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Balancing Performance and Sustainability in Next-Generation PMR Technologies for OMC Structures

Gregory R. Yandek,¹ Jason T. Lamb,² John J. La Scala,³ Benjamin G. Harvey,⁴ Giuseppe R. Palmese,⁵ William S. Eck,⁶ Joshua M. Sadler,³ Santosh K. Yadav⁵

**Air Force Research Laboratory, Rocket Propulsion Division¹, ERC, Inc.², Edwards AFB, CA
U.S. Army Research Laboratory, Weapons & Materials Research Directorate³, Aberdeen Proving Ground, MD**

**Naval Air Warfare Center, Weapons Division⁴, China Lake, CA
Drexel University, Department of Chemical and Biological Engineering⁵, Philadelphia, PA
U.S. Army Public Health Center⁶, Aberdeen Proving Ground, MD**



26 May 2016



WP-2402: Novel Chemistries for Replacement of Methyleneedianiline in Polyimide Composites

Performers:

- J. La Scala, J. Sadler, Army Research Laboratory; B. Harvey, Naval Airfare Warfare Center; G. Palmese, Drexel U.; G. Yandek, Air Force Research Laboratory; W. Eck, Army Public Health Center

Technology Focus

- Develop sustainable high temperature polyimide resins for high performance DoD applications by systematically exploring polymer/composite properties and toxicity of structural variants of MDA using renewable and petroleum-derived chemicals.

Research Objectives

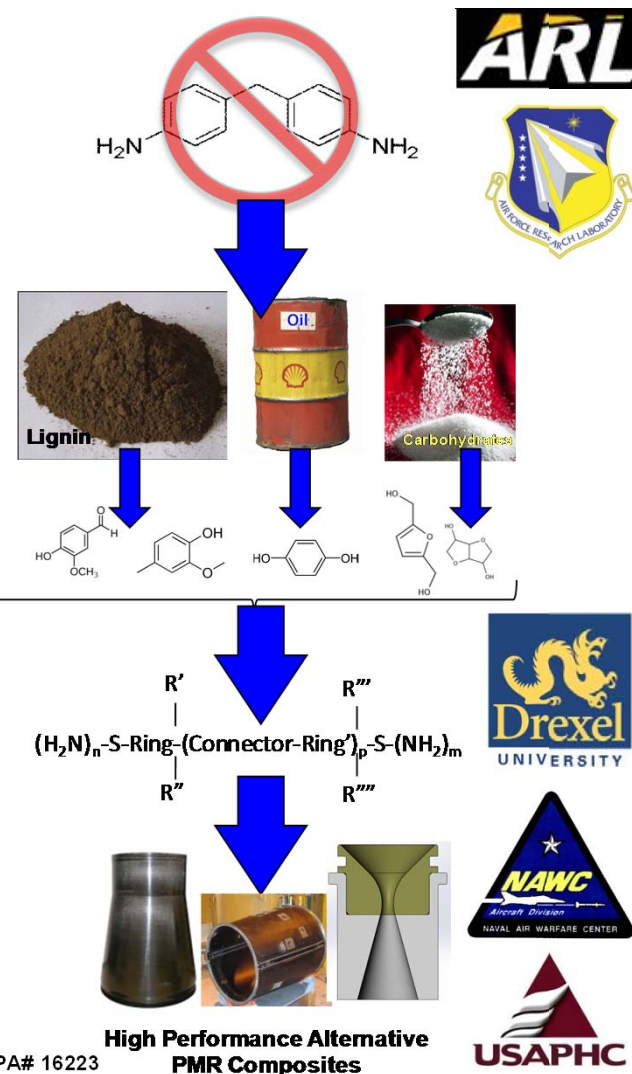
- Develop structure-property-toxicity relationships of MDA alternatives.

Project Progress and Results

- Identified lower toxicity diamines with low toxicity, good processing, and good polymer properties.
- Identified a few gaps in structure-property-toxicity understanding.

Technology Transition

- Transition to Polyimide pre-preg manufacturers and rocket manufacturing primes. Transition to AF and Army high temperature composites programs.





Performers

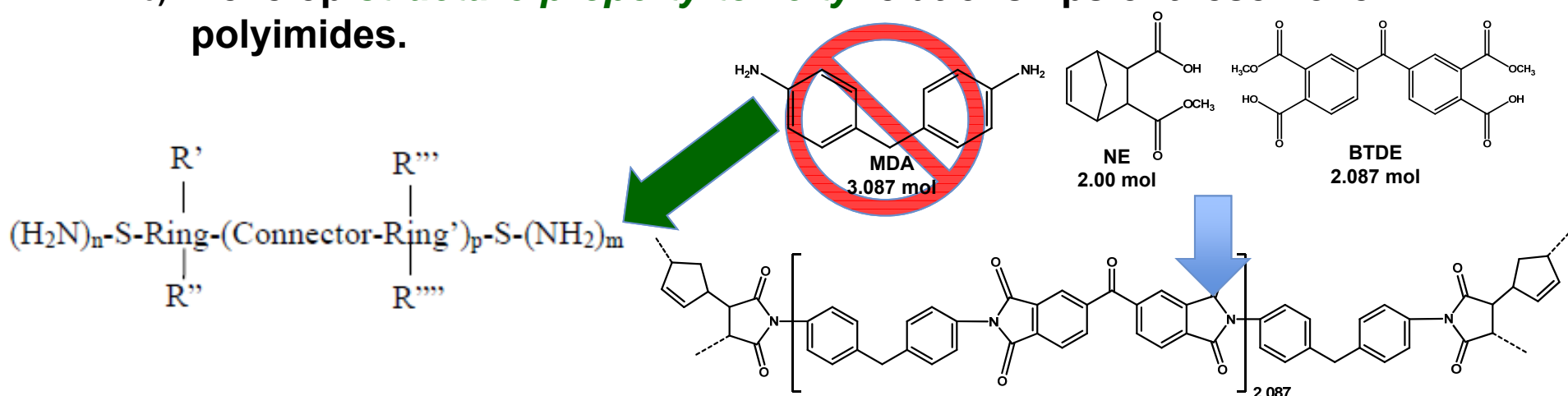
Org.	Team Members	Activities
ARL	Dr. J.J. La Scala (PI) Dr. J. Sadler Dr. C. Paquette Ms. R. Toulan Mr. C. Annunziato Dr. I. McAninch	PI Duties. Chemical preparation and analysis of furan-based and other MDA replacements. Polyimide resin preparation and characterization. Environmental and cost assessment
NAWC	Dr. B. Harvey (co-PI) Dr. M. Garrison Dr. M. Davis Dr. A. Chafin Dr. M. Savolainen	Chemical preparation and analysis of lignin-based and other MDA replacements. Polyimide resin preparation and characterization.
Drexel University	Dr. G.R. Palmese (Co-PI) Mr. F. Hu Dr. S. Yadav	Chemical preparation and analysis of furan-based and other MDA replacements. Polyimide resin preparation and characterization.
AFRL	Dr. G. Yandek (co-PI) Dr. J. Lamb	Pre-preg and composite preparation and characterization.
Army Public Health Center	Dr. W. Eck (co-PI) Dr. V. Adams Dr. C. Cao	QSAR Analysis and Toxicity studies of relevant chemicals and polymers.



Technical Objective

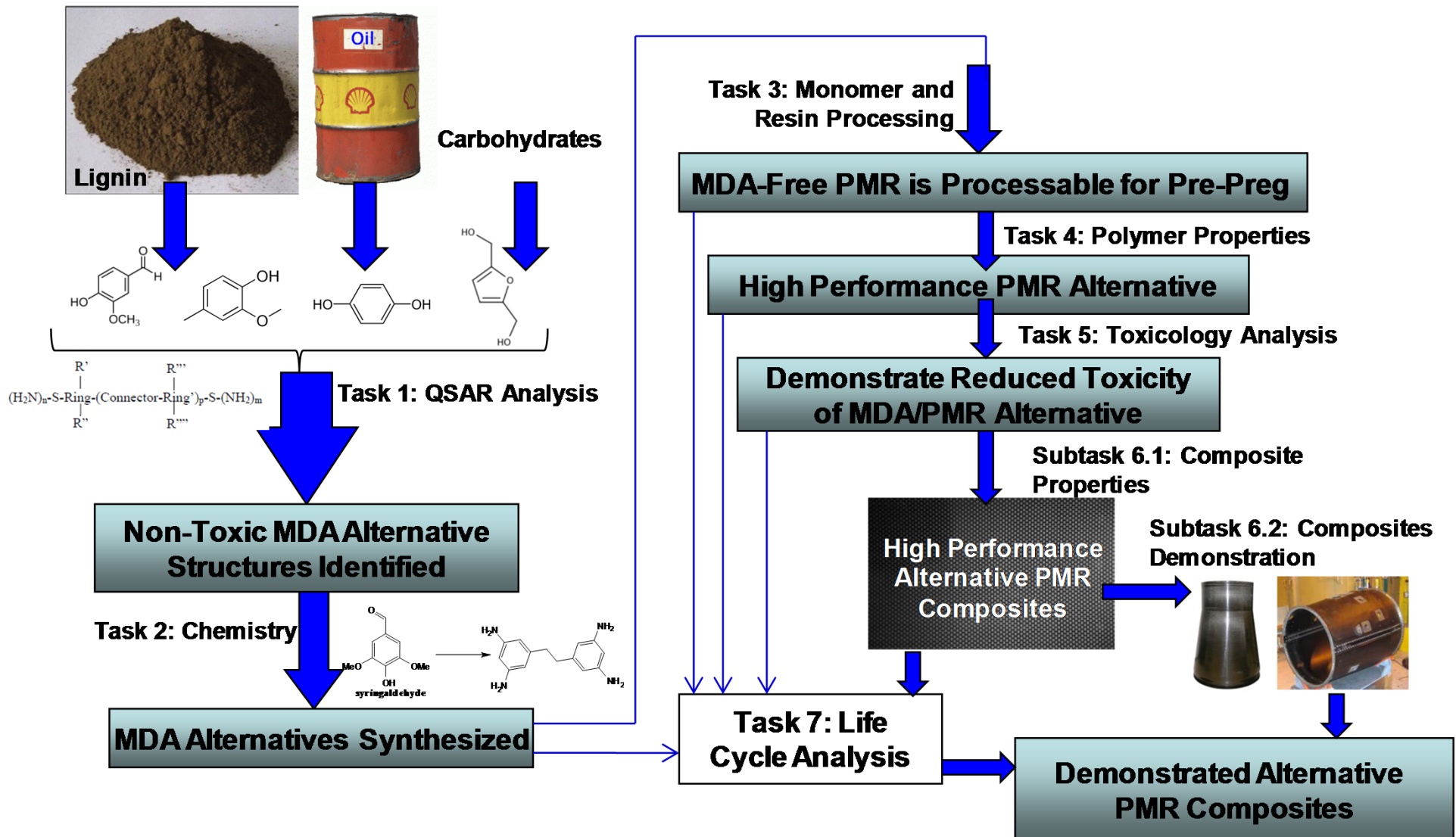
Develop novel chemistries utilizing polymer, health, and environmental sciences to replace MDA in PMR polyimides for high temp composites with both petroleum-derived and renewable/bio-derived monomers that achieve the following:

- 1) Reduce/eliminate toxicity and carcinogenic aspects of the polyimide
- 2) Maintain thermal and mechanical properties of the composites
- 3) Simplify composite processing
- 4) Reduce lifecycle costs
- 5) Develop *structure-property-toxicity* relationships of these novel polyimides.





Technical Approach

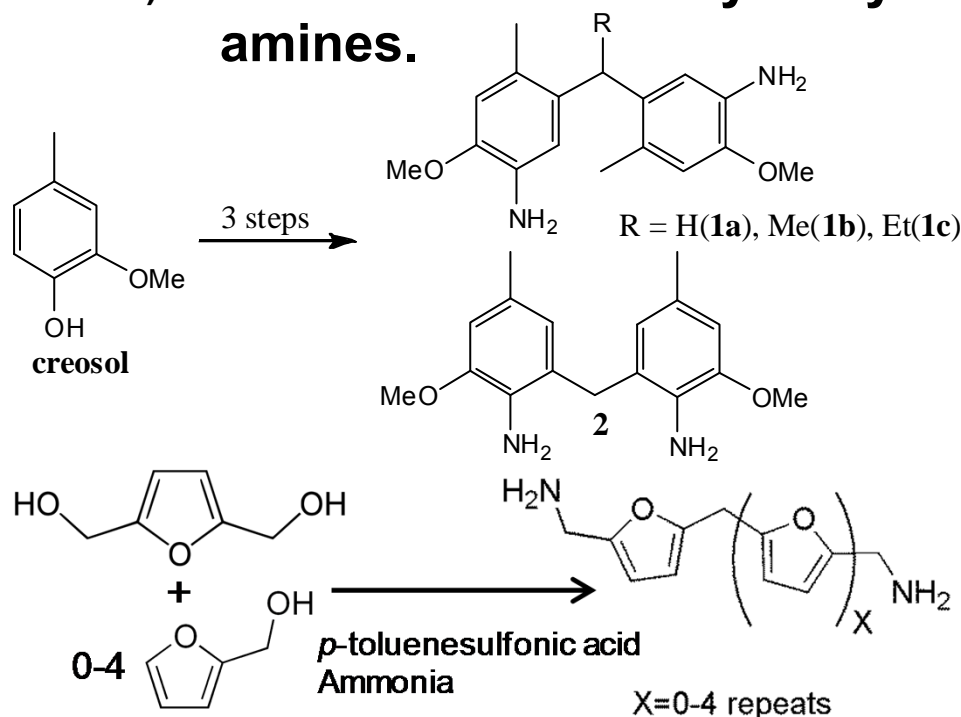




Chemistry to Prepare MDA Alternatives

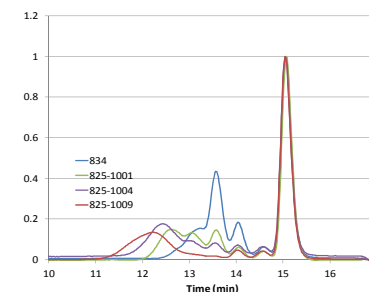
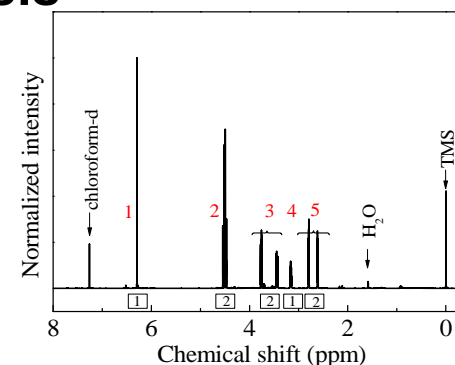
- Various potential rxn schemes.
- Analytical chemistry of products

- 1) Joining of two hydroxyl containing chemicals
- 2) Conversion of the hydroxyls to amines.



Purification using extraction, crystallization, and distillation.

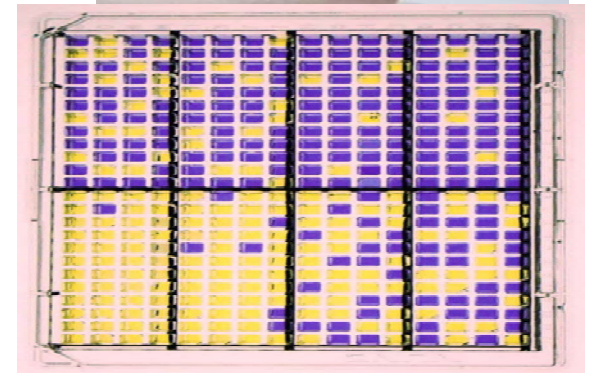
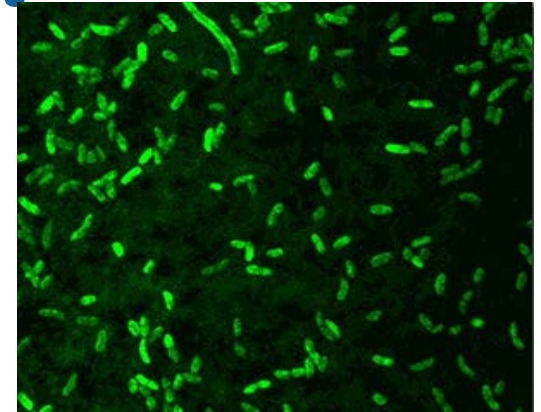
- NMR, FTIR – chemical analysis
- SEC – Molecular weight analysis





Toxicology Assessment

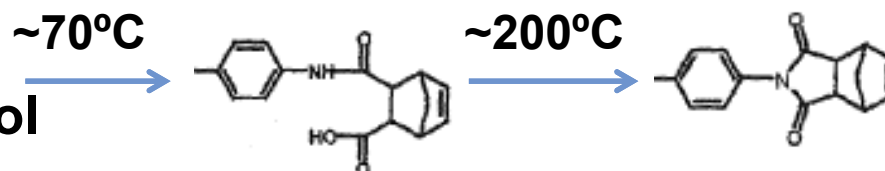
- **Quantitative Structure-Activity Relationship (QSAR) principle** –predict properties based on chemical structure
 - EPA's EPI Suite model package – physical properties
 - TOPKAT – human toxicity prediction
 - ECOTOX modeling system – Environmental toxicity
- ***in vitro* testing**
 - Performed on bacterial cells (e.g. *S. typhimurium* and *E. coli*) with possible follow-on testing in mammalian cells.
 - Test for mutagenicity and aquatic toxicity.
- ***in vivo* testing for high potential MDA alt.**
 - Performed in one species, usually rats.
 - Acute and chronic oral or inhalation toxicity testing, dermal and ocular sensitivity testing, mutagenicity, developmental or reproductive testing, and carcinogenicity testing.



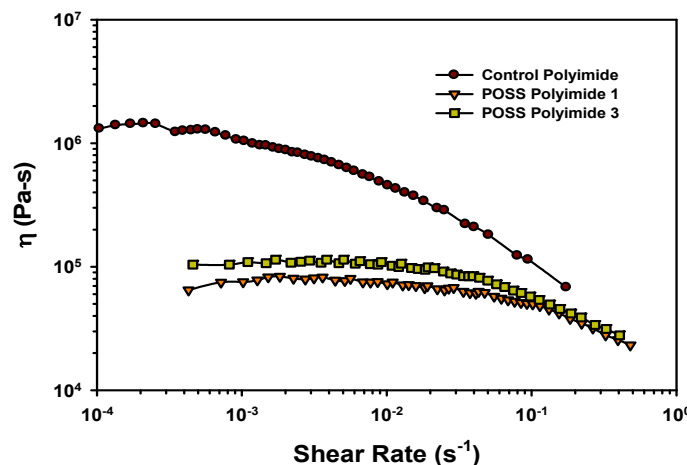


Resin Processing and Cure

- Dissolve NE, BTDE, and MDA alternative in methanol (ethanol as well to replace HAPs).

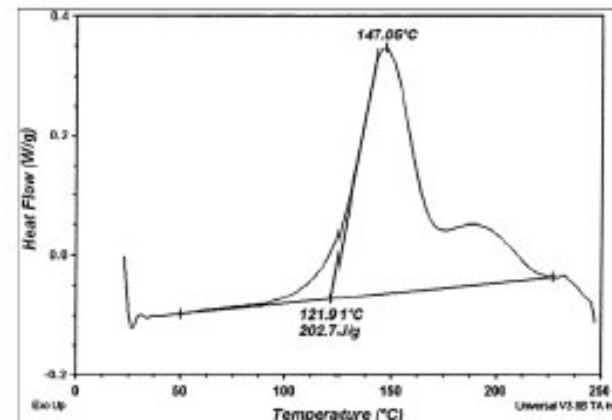


- Assess solubility
- Reactions and analysis
 - Rheology – processing, MW
 - SEC – M_n , M_w , PDI
 - NMR, FTIR – confirm rxns, M_n
- Heats of Reaction



Metrics vs. PMR-15:

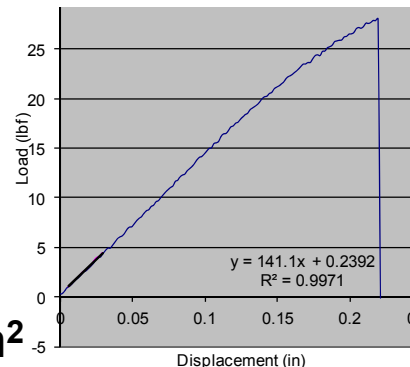
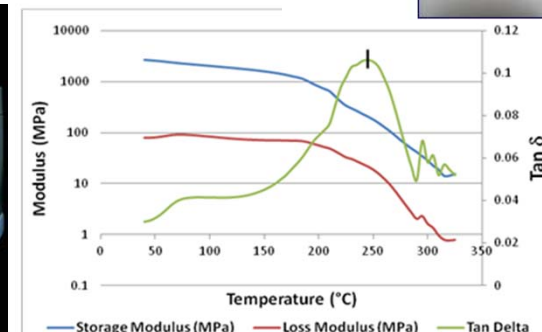
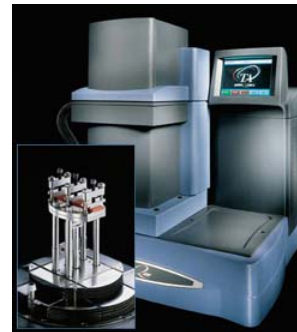
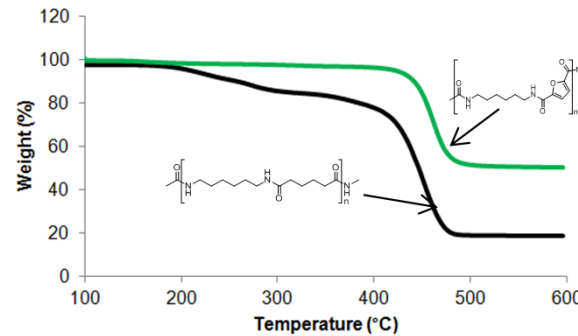
- Similar extent of amic acid olig.
- Imidization extent within 10%
- Amide and imide viscosity within 25%
- Heat of reaction per mole of amine no less than 95%
- Cure temperature no greater +25°C





Neat Resin Properties

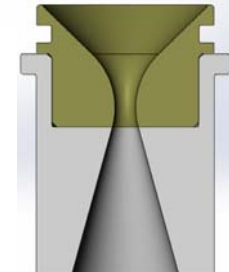
1. Prepare neat samples in steel blocks charged with dried imidized powder under pressure/heat
2. Thermogravimetric analysis to measure degradation temperature (T_d) ($>400^\circ\text{C}$)
3. Dynamic mechanical analysis to measure T_g ($> 95\%$ of 340°C) and crosslink density
4. Thermoxidative stability
 - ~1wt% loss (500 h, 288°C), 5% loss (500 h, 316°C)
 - Retain flexural modulus >3.5 GPa flexural strength >150 MPa (ASTM D638)
5. ASTM D5045 Toughness > 250 J/m²
6. Water absorption (95% RH, 71°C) < 5 wt%





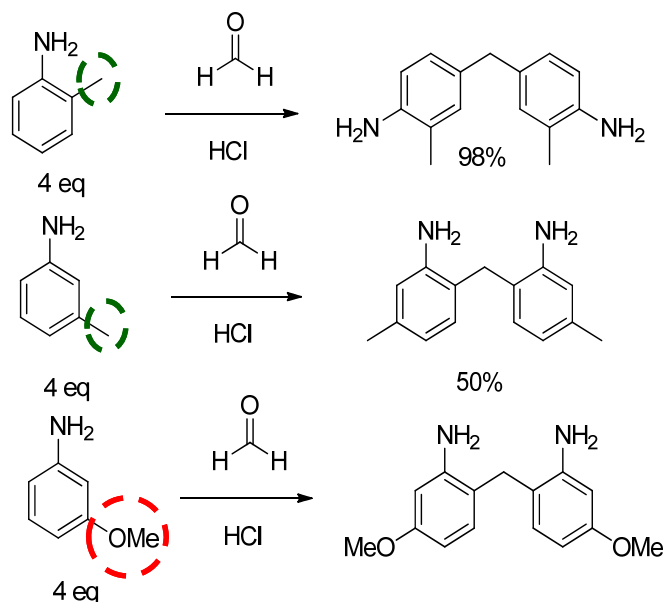
Composites and LCA

- **AFRL to pre-preg carbon and glass fiber with input from Lockheed Martin Skunk Works (LMSW).**
- **Evaluate properties of carbon and glass fiber composites**
 - **Mechanicals**
 - **Water uptake**
 - **Thermo-oxidative exposure**
- **Proof of Concept Parts**
 - **Rocket casings (AFRL) and nozzles (AFRL/AMRDEC)**
- **LCA**
 - **Price of MDA alternatives**
 - **Cost savings due to reduced toxicity**

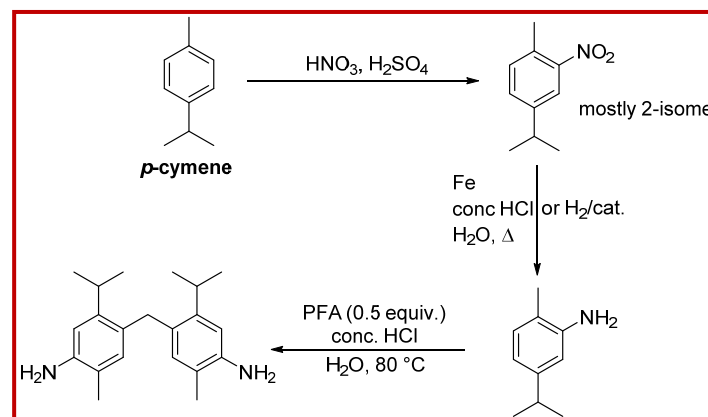




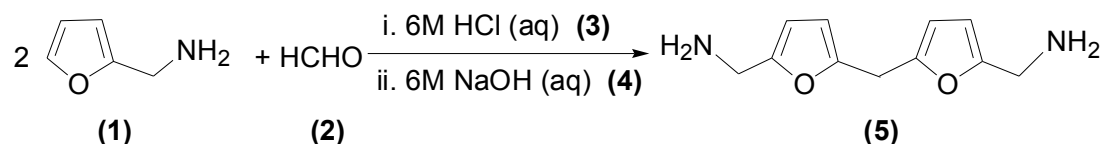
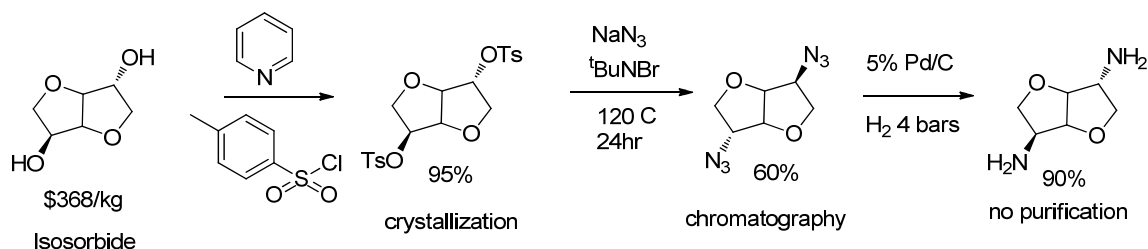
Overview of Prior Work – Synthesis



- Prepared variants of MDA with multiple substituents on the rings



- Prepared variants of MDA with single substituents on the rings



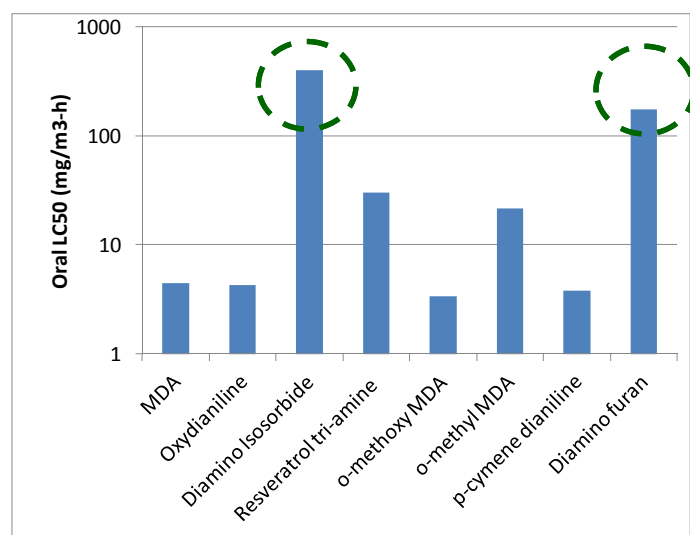
- Prepared MDA variants with different ring chemistry in low yield

- Prepared various potential diamines for structure-property-toxicity studies



Overview of Prior Work – Toxicity

Structure								
Compound	MDA	Oxydianiline	Diamino Isosorbide	Resveratrol tri-amine	o-methoxy MDA	o-methyl MDA	p-cymene dianiline	Diamino furan
Mutagen	Positive (DB value)	Positive (DB value)	Positive (High)	Positive (High)	Positive (High)	Possible (High)	Indeterm	Negative (Low)
Carcinogen	Positive (DB value)	Positive (DB value)	Indeterm	Indeterm	Indeterm	Indeterm	Indeterm	Unlikely



Reduced toxicity

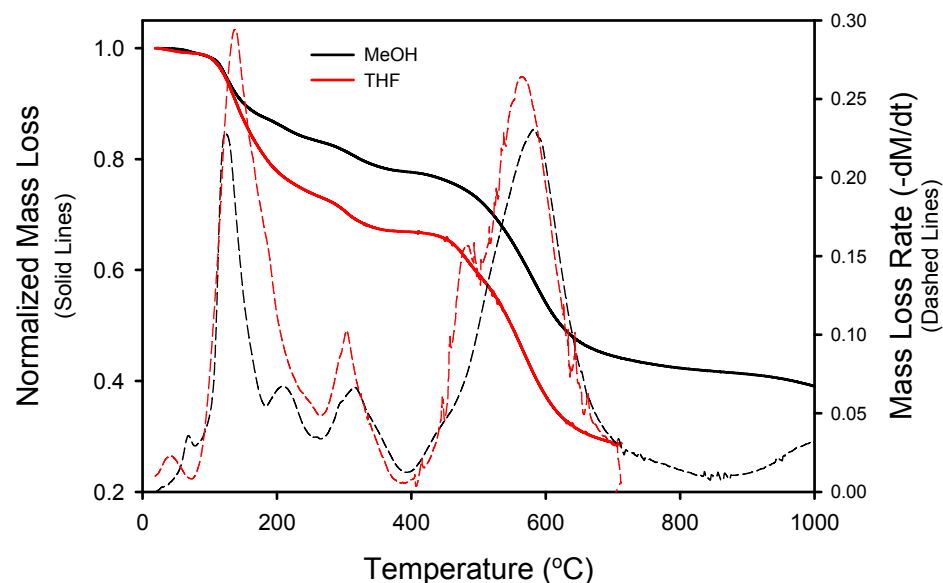


- QSAR results show that some non-phenyl amines have potential for low toxicity
- QSAR modeling completed on 25 compounds



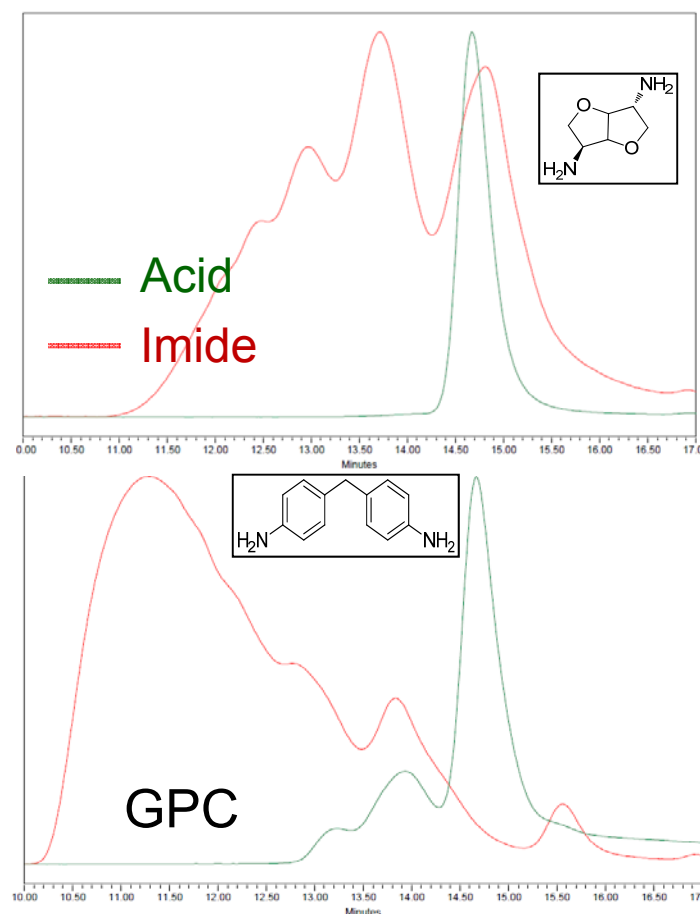
Overview of Prior Work – Processing/Polymerization

- THF synthesis/processing of polyimides investigated to enable use of chemicals not soluble in methanol



- NE incorporated less in polyimide
- Consistently yielded oligomers of higher average MW and PI
- Higher MW oligomers have less crosslinkable endcap / reflected in TGA

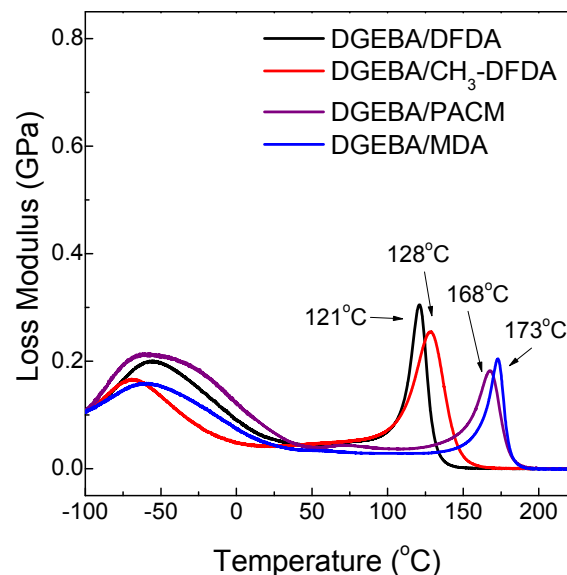
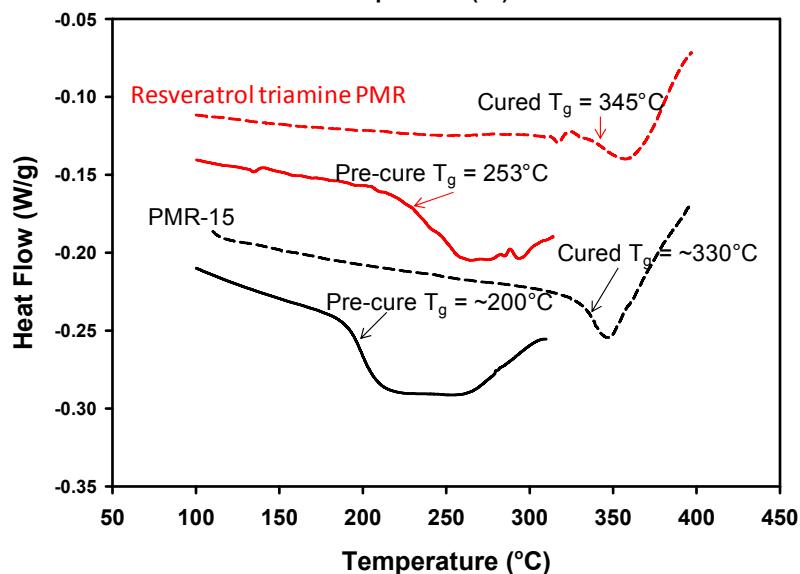
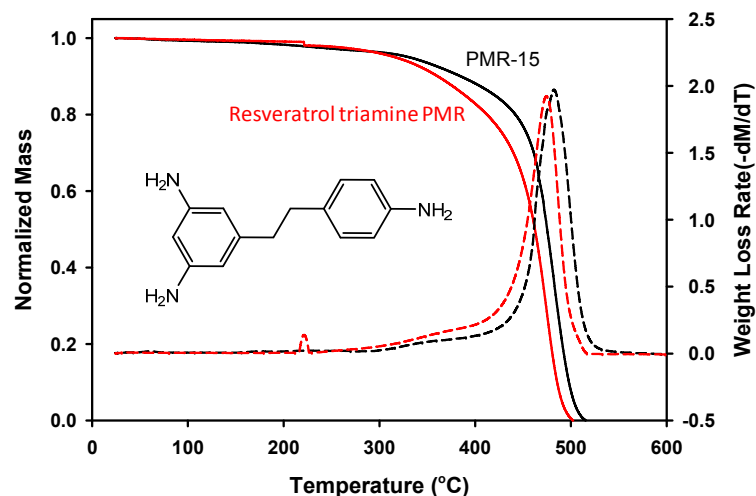
- Monomer choice can affect polyamic acid MW and polyimide MW





Overview of Prior Work – Polymer/Composites Properties

- Validated polymer testing methods
- Resveratrol polyimides with high thermal properties
- Furan-epoxy polymers with high char content but reduced T_g
- High quality polyimide composites (<1% void, 2.5 wt% loss at 316°C for 410 h)





Prior Go/No Go Decisions

- Determine the feasibility of chemically preparing these MDA alternatives
 - ◆ Go – Chemical methods to prepare MDA alternatives are feasible and scalable
- QSARs indicate there are MDA alternatives with reduced toxicity
 - ◆ Go – QSAR suggests a number of compounds with reduced toxicity

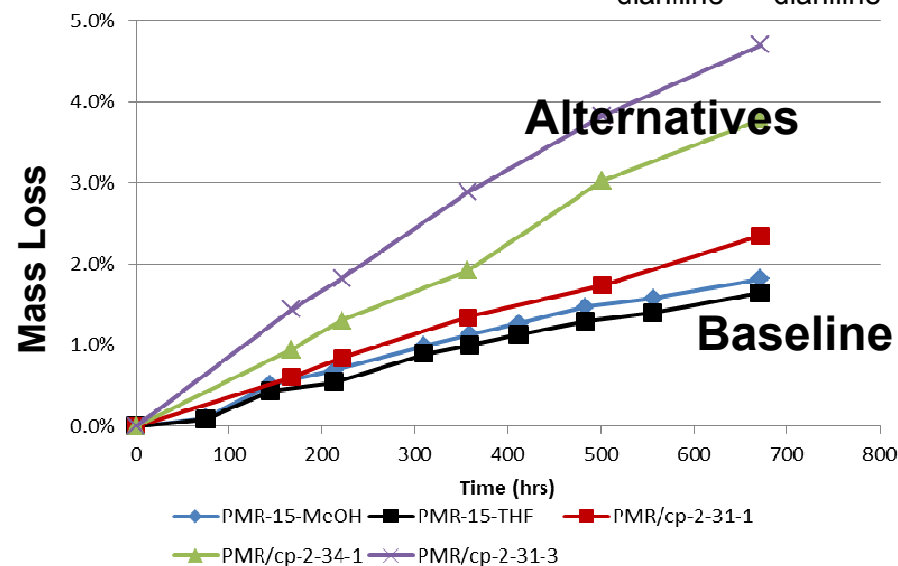
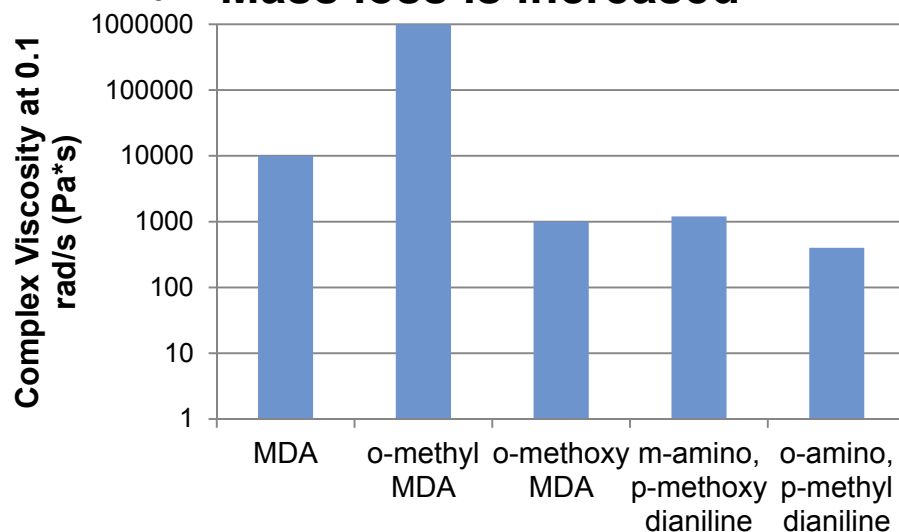
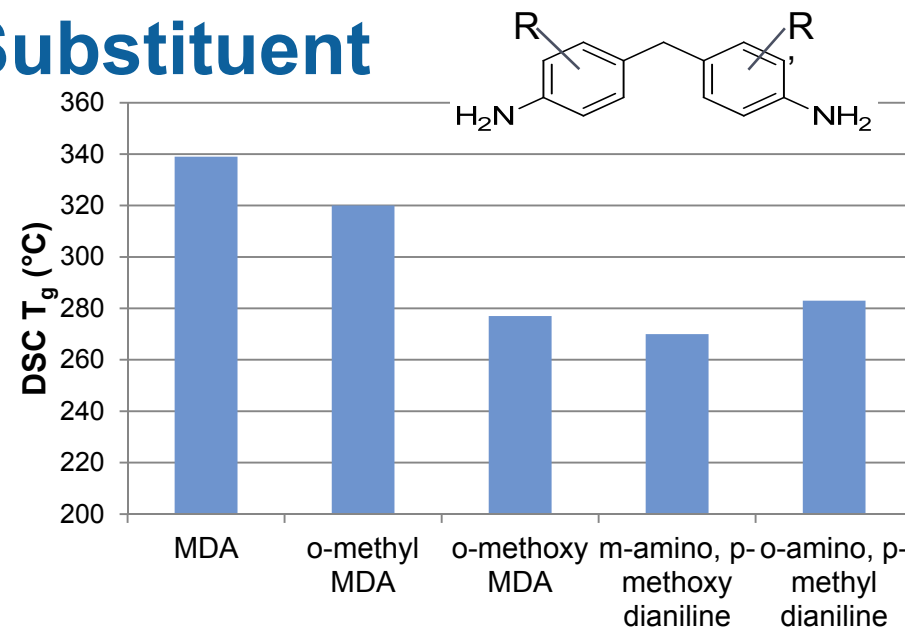


Results – MDA Alternatives with Single Substituent

- Ames Test—**All are mutagenic**

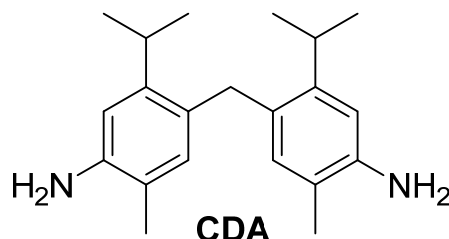
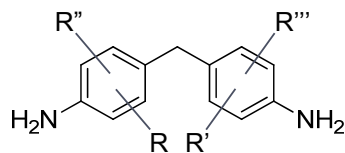
Compound	Microtox (aquatic) mg/L	NRU (acute toxicity) mg/kg
o-methoxy MDA	6.28 (moderate)	781 (very low)
o-amino, p-methyl MDA	42.3 (low)	684 (very low)

- Viscosity varies around MDA
- T_g is generally reduced
- Mass loss is increased





Results – p-Cymene Diamines



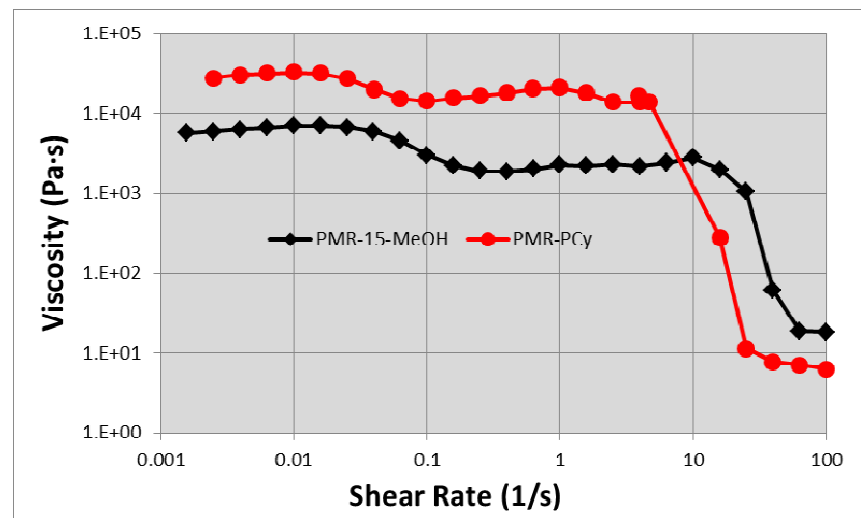
Toxicity

- Ames Test—no mutations for any of the strains tested
- Conclusion: **not mutagenic or carcinogenic**
- LD50: 725 mg/kg (very low tox.)
- Aquatic toxicity EC50: 299.3(1.3) mg/L (non-toxic)
- QSAR shows that all 11 isomers have low oral tox, but have elevated aquatic tox, mitigated by low solubility

Economics

- Turpentine produced at ~350 kton/y at \$0.15/lb
- Limonene produced at 85 kton/y
- \$0.4-3.0/lb on Alibaba for p-cymene
- Should eventually be low cost

Processing



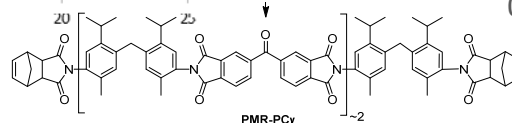
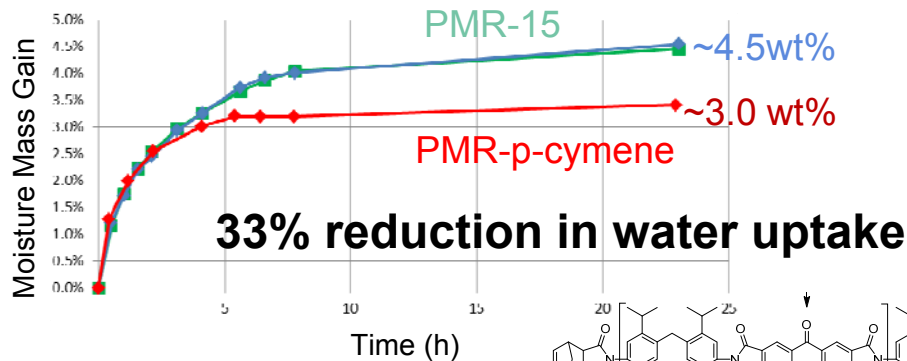
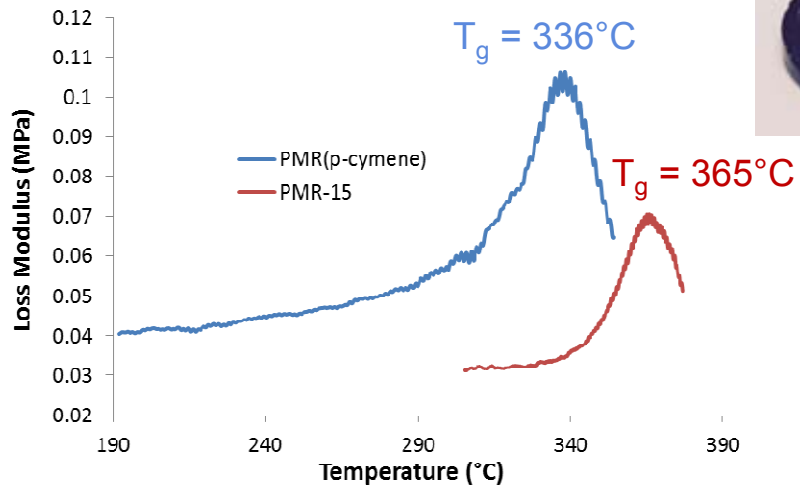
- PMR-PCy has similar viscosity/processing conditions relative to PMR-15
- Current efforts are focused on fabricating flat panels with PMR-PCy using cure protocols and processing parameters developed for PMR-15.



Results – p-Cymene Polyimides and Epoxies

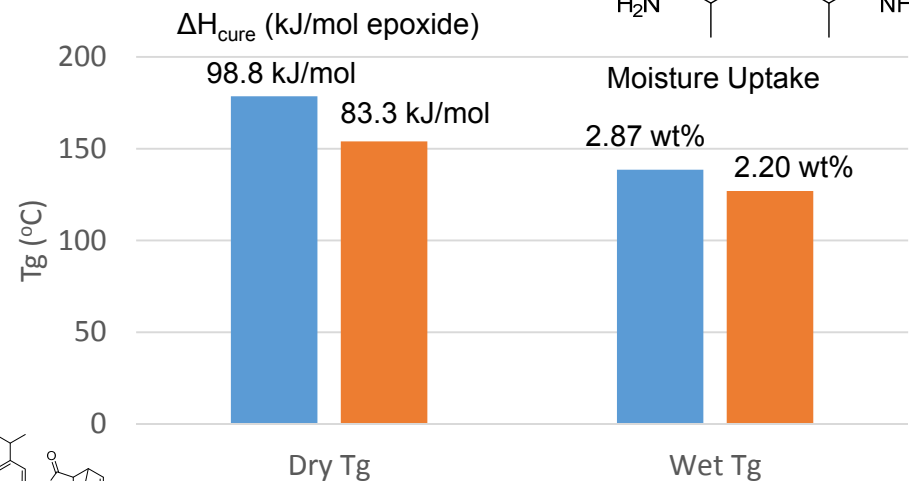
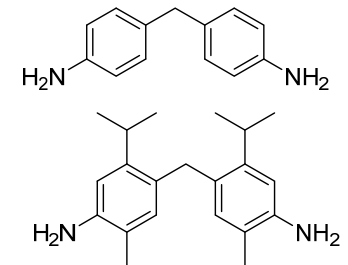
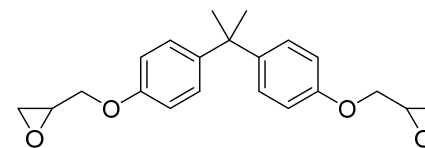
PMR Polyimides

- Lower T_g observed for PMR-PCy via both DSC and TMA (loss peak), but within 0.5-4.5% of PMR-15



Epoxy-Amines

- Methyl and isopropyl groups on amine result in 18-24°C drop in T_g (DSC and TMA) likely due to reduced extent of cure
- 0.4-0.6 wt% (23%) reduction in water uptake
- Similar T_d (~370°C)

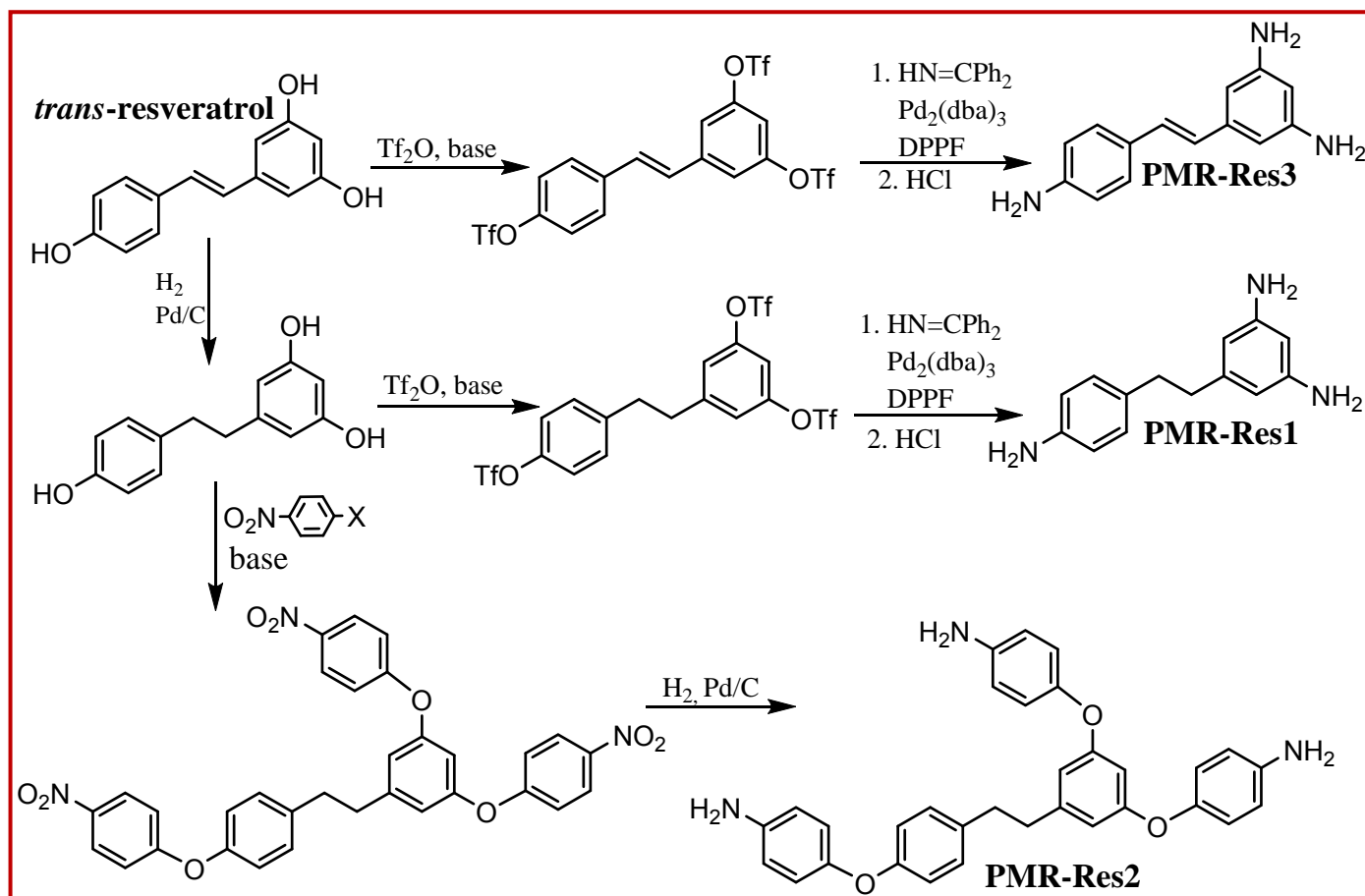


MDA p-cymene



Results – Tris(Aniline) Monomers

- Resveratrol can be generated biosynthetically from sugars.
- Starting trisphenol is conducted in water with low energy demands.
- No formaldehyde or other toxic chemicals are used in the process.



Mutagenic
Cytotoxic $\geq 3.2 \mu\text{g/mL}$

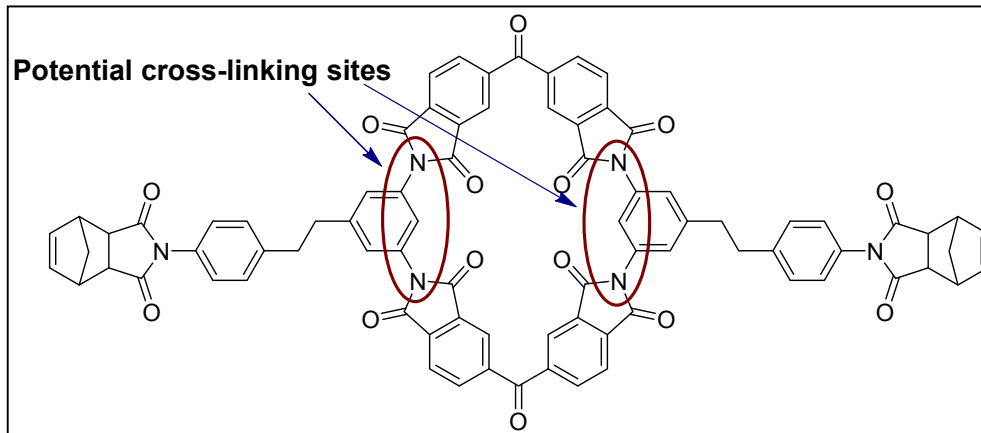
Mutagenic
Cytotoxic $\geq 3.2 \mu\text{g/mL}$

Not Mutagenic based
on AMES testing
No cytotoxicity
observes up to
solubility limit of 10
 $\mu\text{g/mL}$

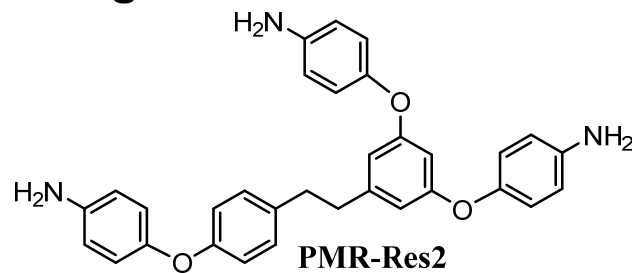


Results – Tris(Aniline) Polyimides

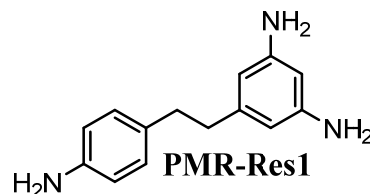
Processing



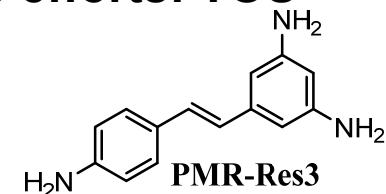
- Trisanilines can cross-link during polyamic acid formation leading to insoluble oligomers.
- Current efforts: Optimize rxn time during formation of the amic acids



Not optimized $T_g = 324^\circ\text{C}$ (DSC)

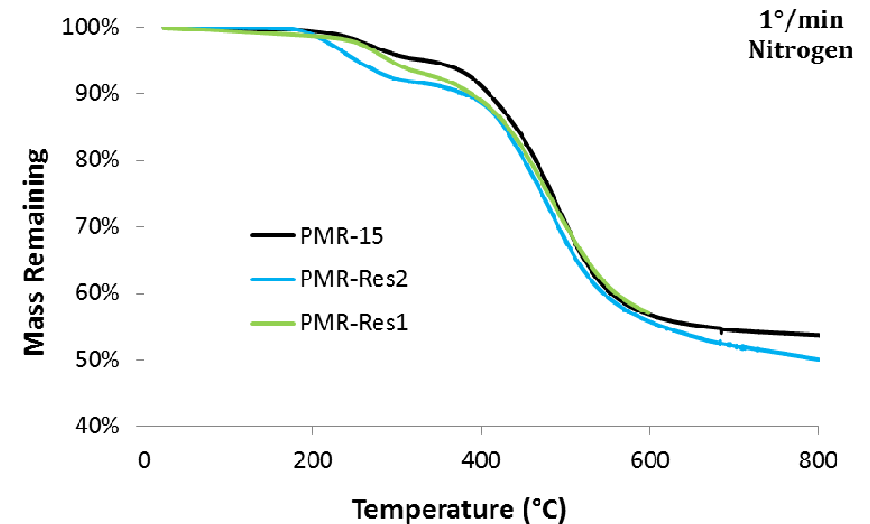


$T_g = 347^\circ\text{C}$ (DSC)



Uncured $T_g = 265^\circ\text{C}$
Postcure $T_g = ???$

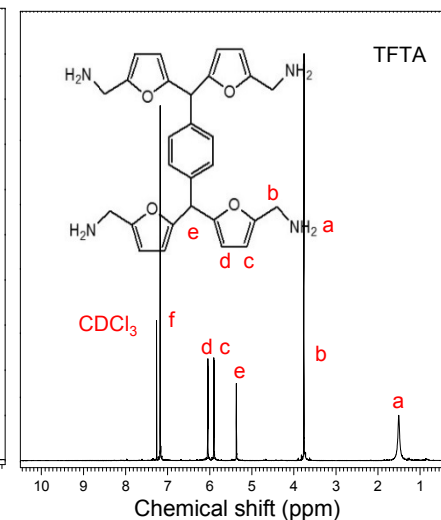
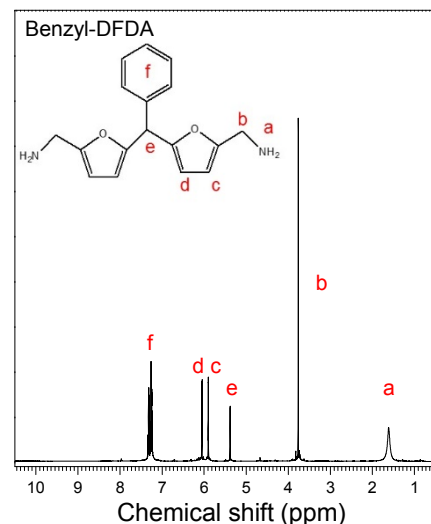
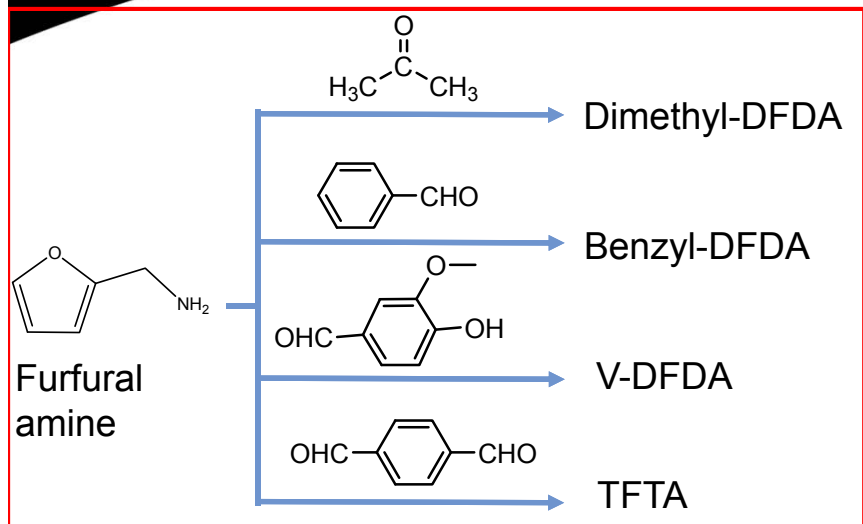
Polymer Properties



- Higher T_g than MDA is possible
- Similar thermal degradation results (TGA)
- Future efforts: TOS



Results: Furan-Based Diamine Synthesis



¹H-NMR spectra of furan based amines

Amines	Purity (1H-NMR)	Amine/aldehyde ratio	Temp.	Time
DFDA	99%	1:2	25 °C	70 min
CH ₃ -DFDA	98%	1:2	40 °C	70 min
DM-DFDA	97%	1:2	40 °C	24 h
Benzyl-DFDA	99%	1:10	40 °C	24 h
TFTA	99%	1:20	40 °C	24 h
V-DFDA	99%	1:10	40 °C	24 h

Toxicity

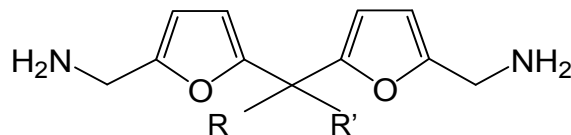
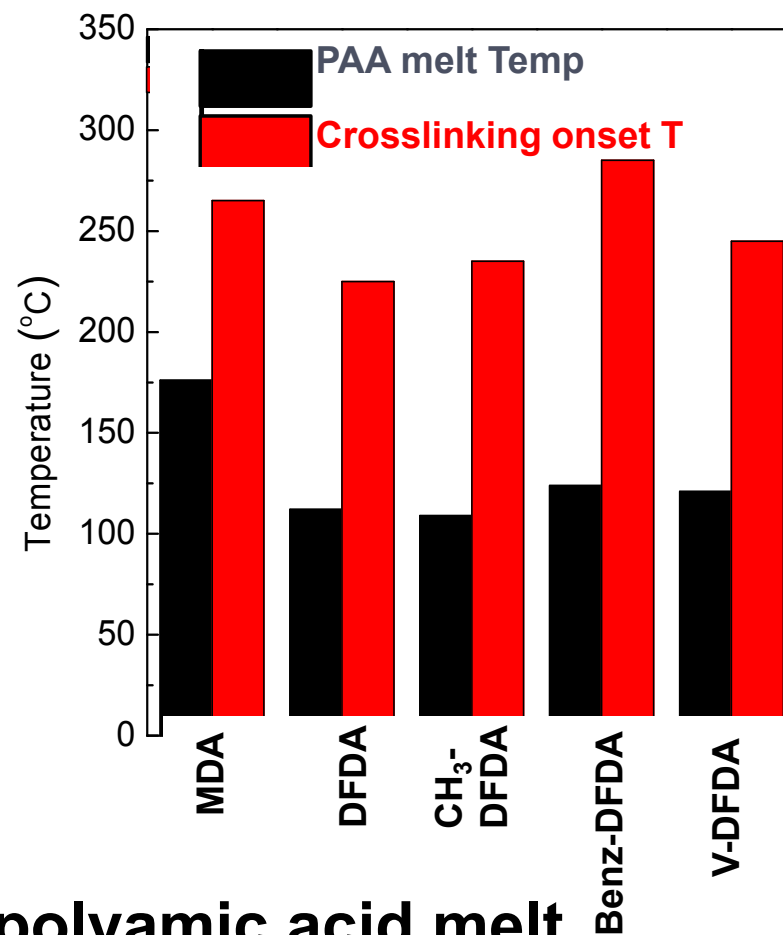
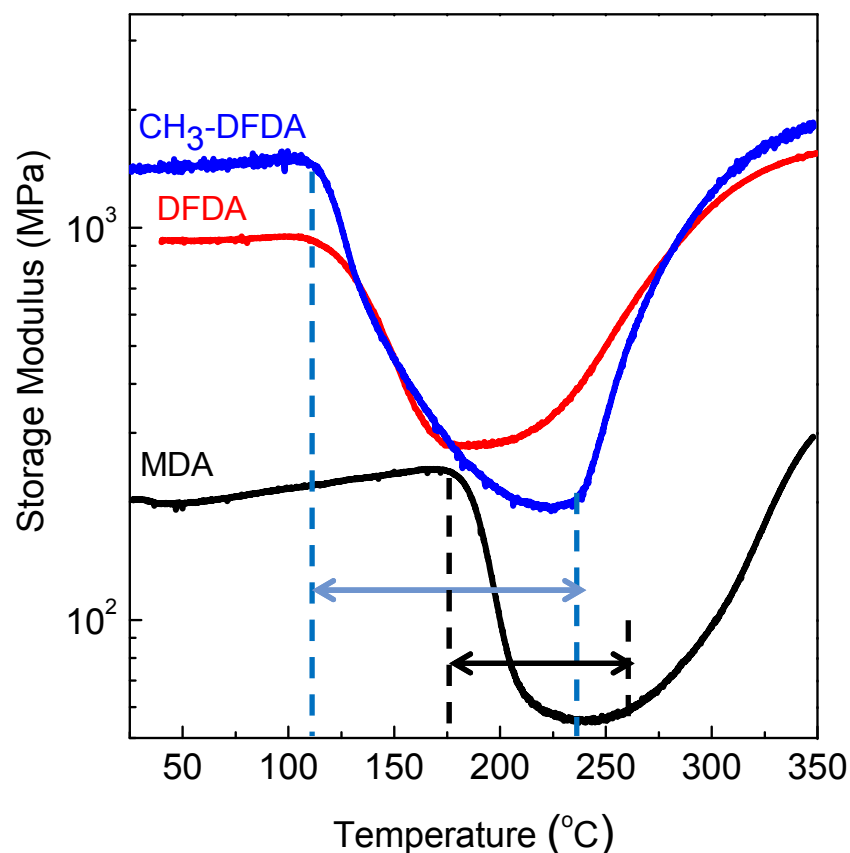
- **QSAR – predicted low toxicity for all R and R'**
- **Not Mutagenic based on AMES testing**
- **No cytotoxicity observed up to solubility limit of 10 µg/mL**

- **One-step** synthesis method with high purity.
- Benzyl-DFDA, V-DFDA and TFTA are combinations of phenyl and furfural building blocks.
- A wide variety of bio-based aldehydes can be used to prepare furan based amines.

Result: Prepared various novel furan based diamines for PMR synthesis.



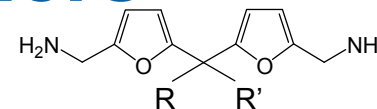
Results – Furan Polyimides Processing



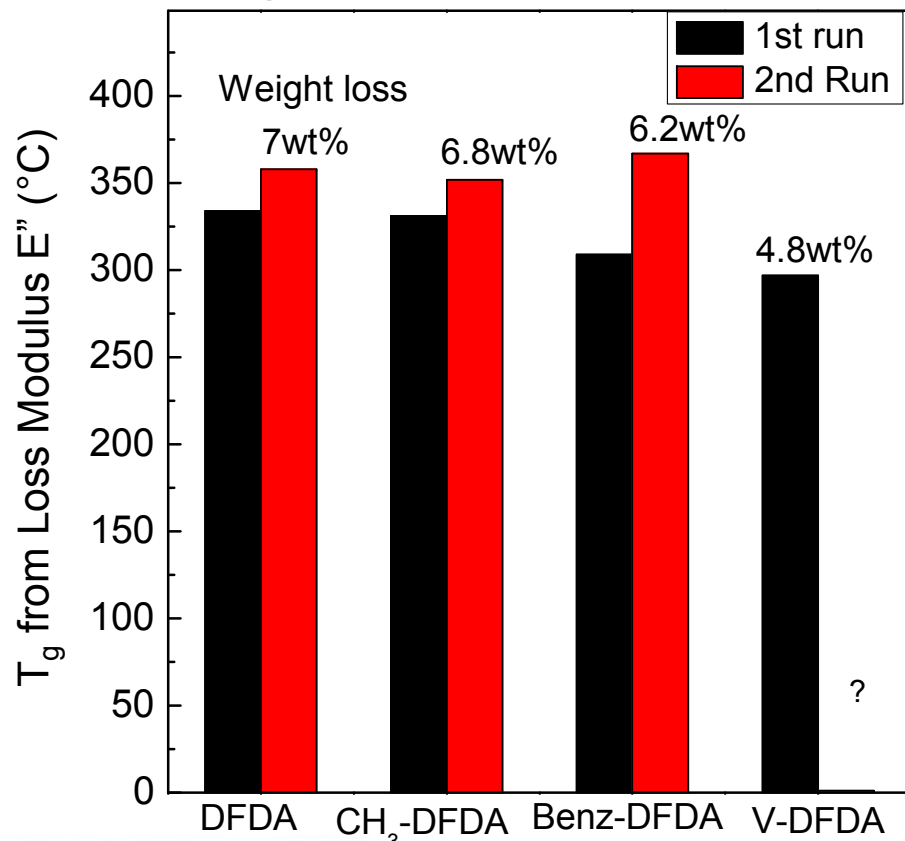
- Lower polyamic acid melt temperature for furans
- Lower curing temperature for furans



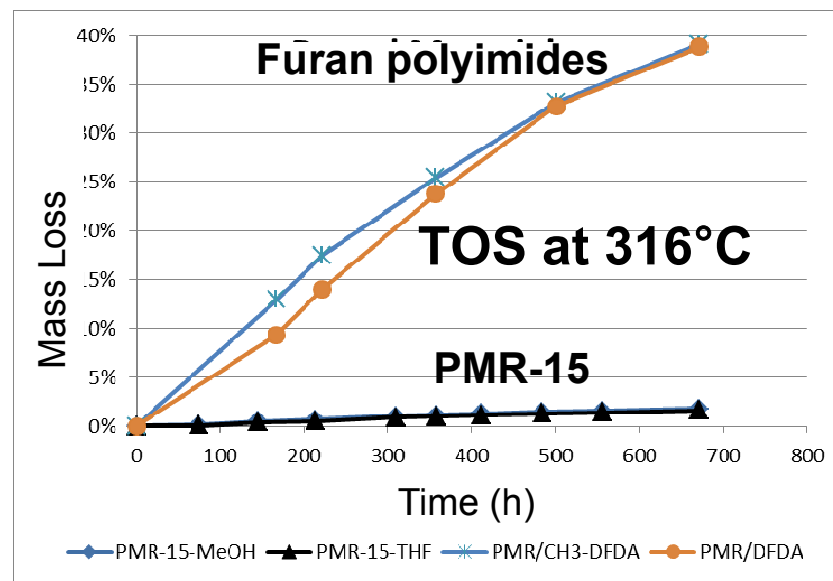
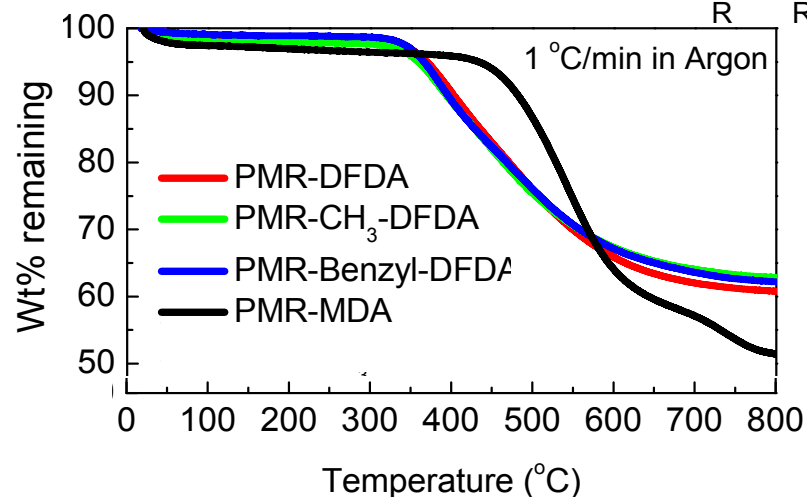
Results – Furan Polyimide Polymers



$T_g(E'') = 365^\circ\text{C}$ for PMR-15



- Furan polyimides can have high T_g
- TOS is an issue at very high temperatures

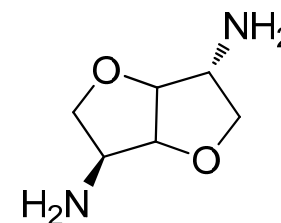




Results – Diamino Isosorbide

Toxicity

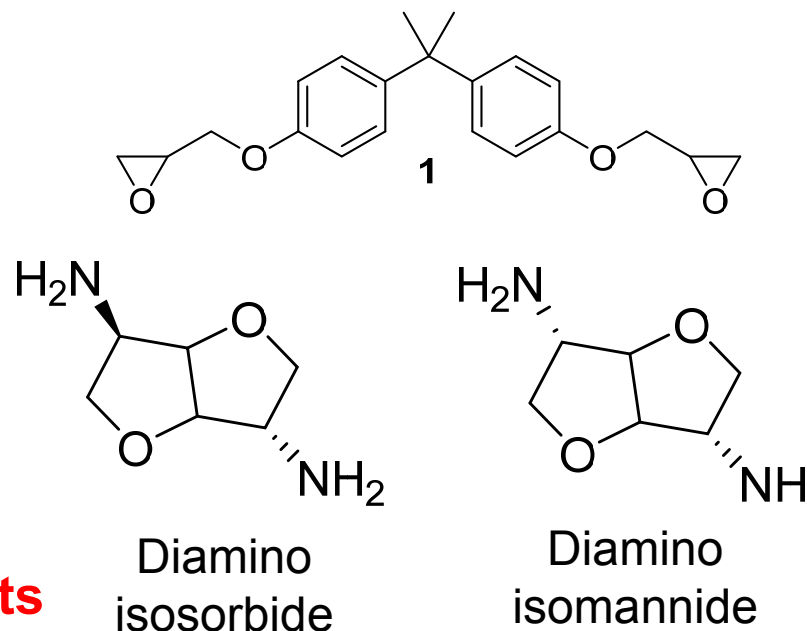
- Ames Test—no mutations for any of the strains tested
- Conclusion: **not mutagenic or carcinogenic**
- LD50: >2000 mg/kg (not toxic)
- Aquatic toxicity EC50: 19.84(1.3) mg/L (low-toxicity)



- Previous results showed poor polyamic acid formation
- Polyimide $T_g = 240^\circ \text{C}$

Diamine	ΔH_{rxn} (kJ/mol)	Normalized ΔH_{rxn}
MDA	373	1
Diamino isosorbide	0	0
Diamino isomannide	315	0.84

- **Stereochemistry of isosorbide prevents its utility as a diamine curing agent**



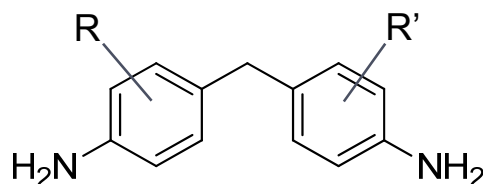


Results – Structure-Property-Toxicity

Positives

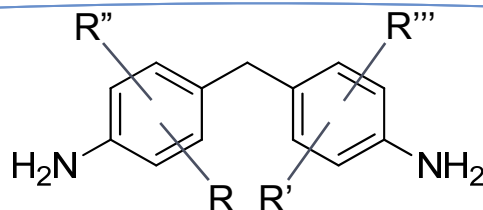
Negatives

- Inexpensive reactants

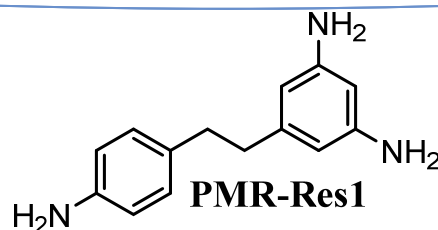


- High toxicity

- Low toxicity
- High thermal properties

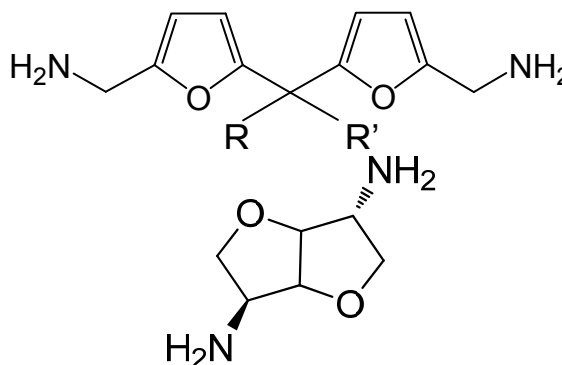


- Can have improved thermal properties



- More difficulty in processing (crosslinking)
- Relatively expensive reactants (currently)

- Low toxicity
- Improved processing
- Reasonable thermal properties

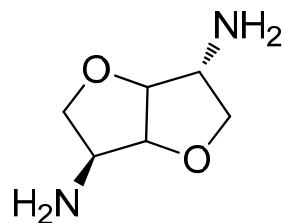
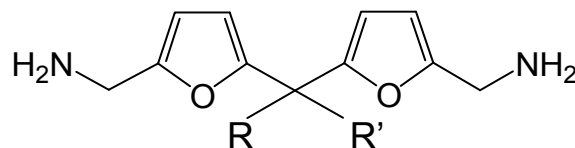
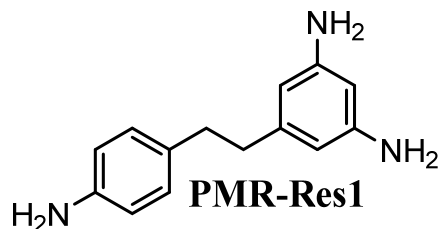
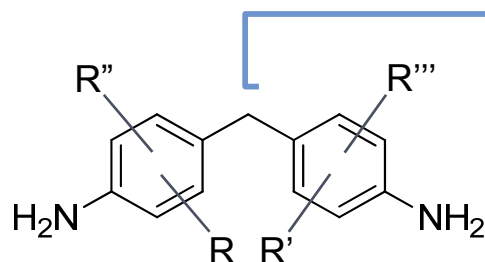
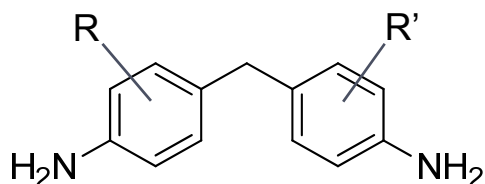


- Reduced TOS
- Low polymer properties (poor reactivity)

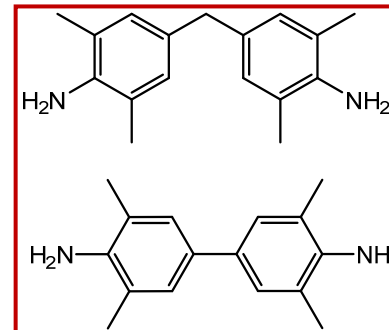
- Low toxicity



Gaps/Work to Complete



- No go as MDA alternative
- Polyimide and epoxy papers
- Do two substituents reduce toxicity or do two specific substituents reduce toxicity?
 - ♦ Additional monomers/polymers to address this gap
- Polyimide papers
- Polyimide composites
- Optimize polyamic acid formation
- Polyimide composites only using PMR-Res2
- TOS studies
- Paper on how substituents affect properties
- Polyimide composites for lower temp applications
- No go as MDA alternative
- Very low priority – identify catalyst to enable facile polyamic acid and polyimide formation
- Report in above papers



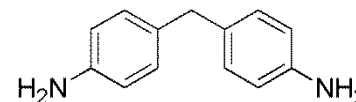


Summary and Conclusions

- Identified a number of non-toxic diamines.
- QSAR indicates that all diamines evaluated had minimal endocrine disruption. (H. Hong et al., FDA leveraging)
- Multi-substituents on phenolic ring results in low toxicity, good processing, and high properties.
- Some gaps remain in structure-property-toxicity studies
- Move forward with scale-up and composites evaluations of promising formulations



Problem Statement



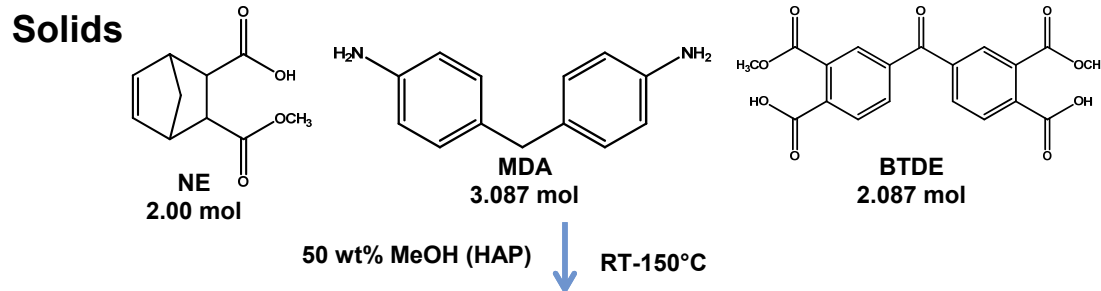
- **4,4'-methylene dianiline (MDA) is a liver toxin and carcinogen.**
- **PMR-15 uses MDA (Monsanto Skybond 701 and LARC-Si as well)**
 - ◆ Usage: ~20,000 lbs/yr PMR-15 – almost all for DoD applications.
 - ◆ Aircraft engines, rocket casings, engine bypass ducts, nozzle flaps, bushings, bearings, missile fins, structural wing components, radar domes, and other DoD applications requiring high temperature composites.
 - ◆ Largest use: Outer Bypass Duct of GE Aviation's F/A-18E/F & EA-18G engines.
 - ◆ Most versatile polyimide – Excellent balance of processing, mechanical, and thermal properties
- Toxicity poses added cost to government in increased production costs and decreased competition.
- EPA Composite Manufacturing and Repair National Emission Standard for Hazardous Air Pollutants regulations, Army Environmental Requirements and Technology Assessments, and REACH **prohibit new use.**



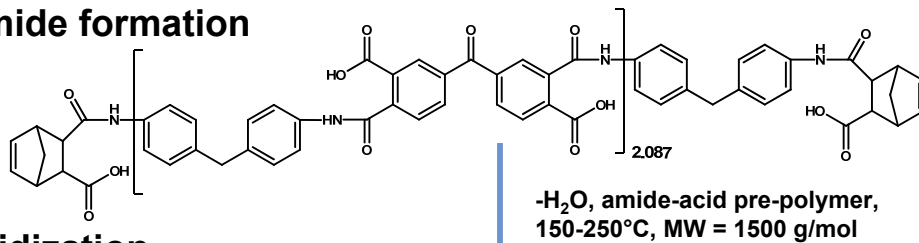


Technical Background – PMR-15

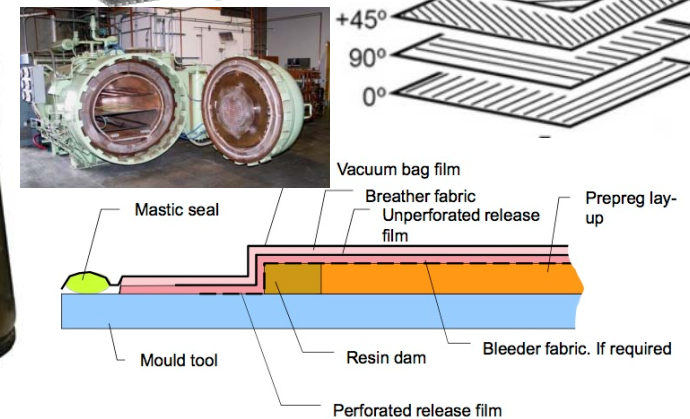
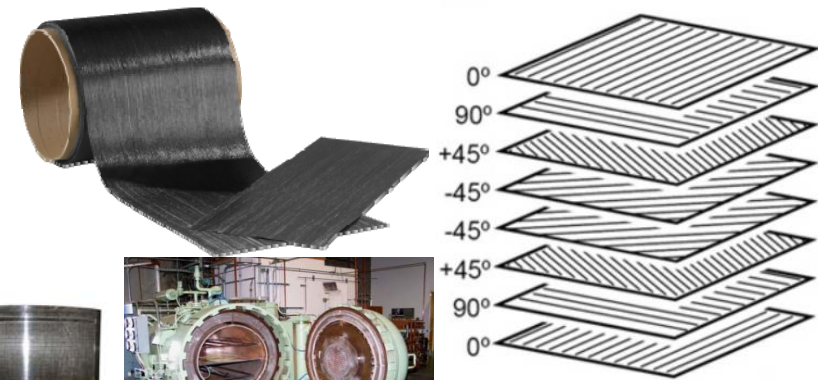
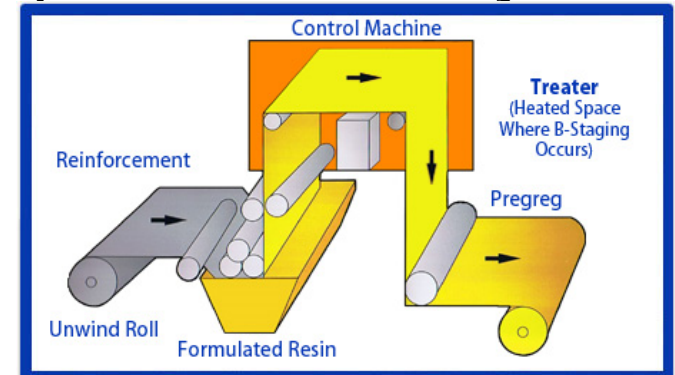
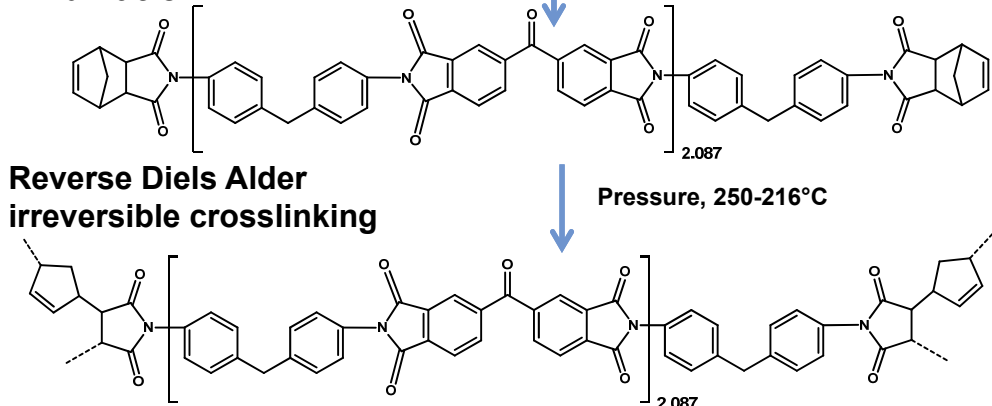
- PMR-15 is Polymerization of the *Monomer Reactants* MDA, 5-norbornene-2,3'-dicarboxylic half acid ester (NE), and 3,3',4,4'-benzophenonetetracarboxylic diester (BTDE)



Amide formation



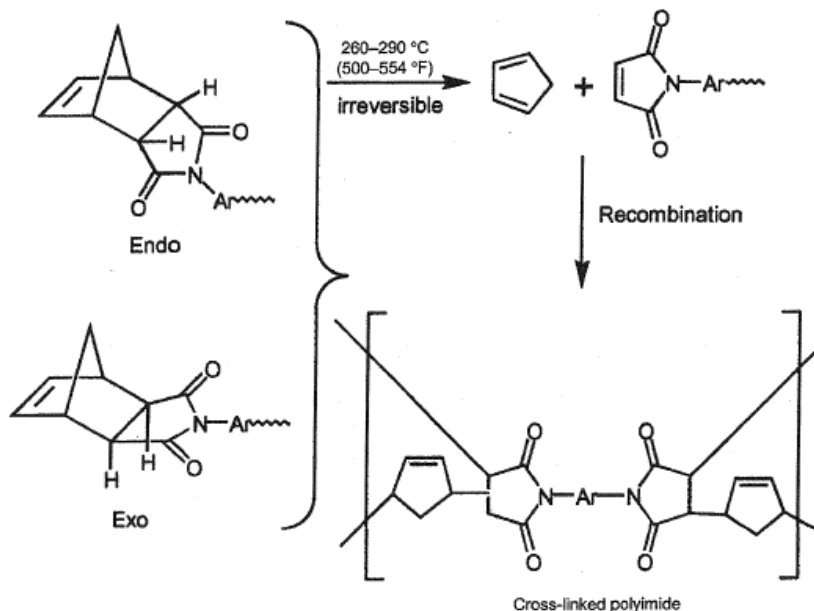
Reverse Diels Alder irreversible crosslinking



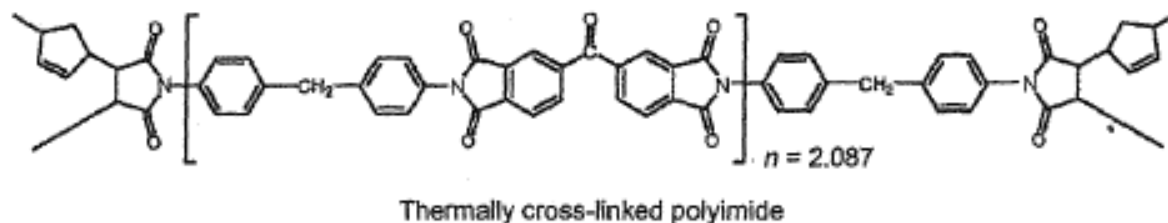


Technical Background – PMR-15

- Reverse Diels Alder irreversible crosslinking



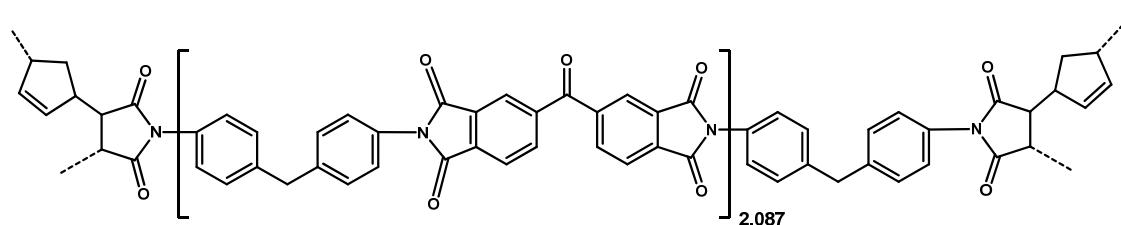
- Highly aromatic, sp^2 planar bonded structure
 - Results in high glass transition temperature and degradation temperature
- Exceptions to sp^2 structure:
 - Methylene bridge between the phenyl units in MDA and the crosslinking of the NE
 - Provides toughness





Technical Background – Properties

- Highly aromatic, sp^2 planar bonded structure results in high glass transition temperature (T_g) and degradation temperature (T_d)
- Exceptions to sp^2 structure provide toughness and improved processability



Neat resin properties

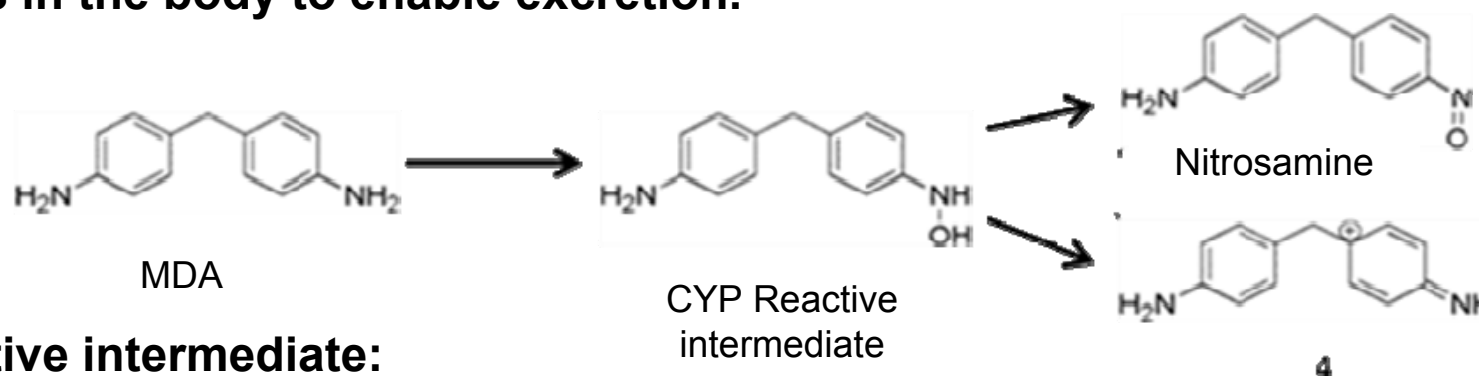
Property	Modulus (GPa)	Strength (MPa)	Elongation (%)
Flexural	4.0	~175	1.5
Tensile	3.9	~40	1.1

- Fracture toughness $\sim 280 \text{ J/m}^2$
- Water absorption at 95% RH and $71^\circ\text{C} = 4.2\%$
- Other polyimides can have higher glass transition temperature (T_g), strength, toughness, OR degradation temperature, but PMR-15 has highest aggregate of these properties.
- $T_g = 340^\circ\text{C}$ (320°C after 1 h at 316°C)
- Weight loss under flowing air:
 - $\sim 1\%$ (500 h, 288°C)
 - $\sim 5\%$ (500 h, 316°C)

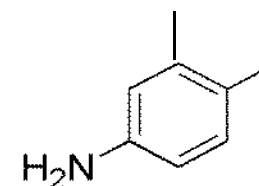


Technical Background – MDA Toxicity

- Liver uses cytochrome P450 (CYP) enzyme to oxidize organic chemicals in the body to enable excretion.



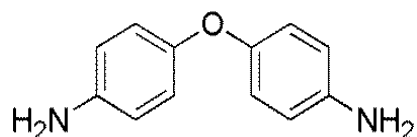
- CYP reactive intermediate:
 - Is highly reactive and damages liver tissue.
 - Readily yields carcinogenic nitrosamine.
- A.G. Siraki, et al., “N-oxidation of aromatic amines by intracellular oxidases.” *Drug Metab Rev.*, 2002, 34, 549-564.
 - The reactivity of CYP towards a chemical is based on the chemical's functional groups, shape, and size.
- Changes in the electronics of the aromatic ring should yield derivatives with lower reactivity towards metabolizing enzymes resulting and affect solubility in reduced toxicity.





Technical Background – MDA Alternatives

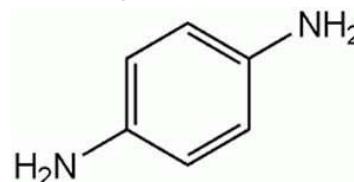
- Other diamine monomers have been studied and used.



4,4'-oxydianiline

Kapton, LARC PDMDA, Amoco
Ultradel 4212, Mitsui-Tuatso TPI
Aurum, 6F Pyralin, LARC RP-46

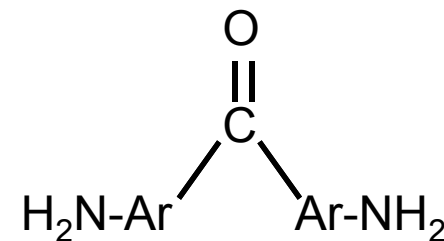
Phenylene diamine



**AFR 700B, VCAP-II,
Superimide 800**



**Amoco Ultradel 4212,
Mitsui-Tuatso TPI Aurum**



LARC-TPI

- 4,4'-oxydianiline reduces toxicity (but is still toxic), but also thermal stability.** R.H. Pater, "Thermosetting polyimides: a review." *SAMPE J.*, 1994, **30**, 29–38.
- Other alternatives are just as toxic as MDA (but used less) and affect polymer properties.**



Technical Background

MDA Alternatives – Technical Gaps

- **No systematic approach to monomer architecture and its effect on toxicity.**
- **No consideration of non-phenyl aromatic groups, such as furans, and non-aromatic cyclic groups, such as isosorbide, as units of the PMR monomers to reduce toxicity and rigidize the resulting polymer.**



Toxicity Assessment

- **ESOH toxicity assessment performed in accordance with ASTM Guide E2552 and USAPHC Standing Operating Procedures.**
- **Toxicity testing conducted in a phased manner, according to the level of item development.**
 - ◆ **Quicker, less expensive testing done initially**
 - ◆ **More definitive, costly testing done on best candidate(s)**